

(19)



Europäisches Patentamt
European Patent Office
Office européen des brevets

(11) Publication number:

**0 272 025
A2**

(12)

EUROPEAN PATENT APPLICATION

(21) Application number: 87310709.8

(51) Int. Cl.4: **C12N 9/80**, **C12P 7/40**,
//**(C12N9/80,C12R1:01)**

(22) Date of filing: 04.12.87

The microorganism(s) has (have) been deposited with NCIB, NRRL and FRI under number(s) NCIB 10508 - 10515, NCIB 10592 - 10596, NCIB 11585, NRRL B5352 - B5364 and FRI 1215-1227.

(71) Applicant: **IMPERIAL CHEMICAL INDUSTRIES PLC**
Imperial Chemical House Millbank
London SW1P 3JF(GB)

(30) Priority: 16.12.86 GB 8630029

(72) Inventor: **Byrom, David**
27 Moor Park Nunthorpe
Middlesborough Cleveland TS7 0JJ(GB)
Inventor: **Carver, Mark Andrew Hallswood**
503 Yarm Road Eaglescliffe
Stockton-on-Tees Cleveland TS16 9BE(GB)

(43) Date of publication of application:
22.06.88 Bulletin 88/25(64) Designated Contracting States:
BE DE ES FR GB IT NL

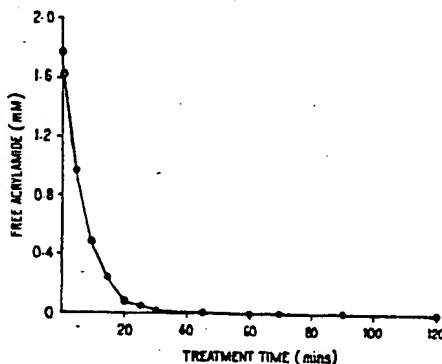
(74) Representative: **Aufflick, James Neil et al**
Imperial Chemical Industries PLC Legal
Department: Patents PO Box 6 Bessemer
Road
Welwyn Garden City Herts, AL7 1HD(GB)

(54) **Process for the decomposition of acrylamide.**

(57) A process for the decomposition of acrylamide using an amidase (acrylamide amido hydrolase) enzyme in which the enzyme is induced in strains of Methylophilus methylotrophus. A method for inducing the enzyme and a process for producing acrylic acid are also claimed. The process for the decomposition of acrylamide is useful in particular for reducing the level of unreacted monomer associated with homo- and hetero-polymers of acrylamide. The activity of the enzyme can be increased by heating it to a temperature in the range 40° to 80°C.

FIG.2

AMIDASE REMOVAL OF ACRYLAMIDE FROM AN ANIONIC
LATEX TYPE A628



ANIONIC LATEX, NALFLOC TYPE A628
ENZYME LOADING 1000 U/kg
TEMPERATURE 60°C
pH 5.9

Xerox Copy Centre

EP 0 272 025 A2

Process for the decomposition of acrylamide

This invention relates to a process for the decomposition of acrylamide, to a method for the production of the enzyme amidase (acrylamide amido hydrolase E.C. No. 3-5-1-4) used in the decomposition of acrylamide and to a process for the production of acrylic acid or salt or ester thereof by decomposition of acrylamide.

5 Polyacrylamide polymers, i.e. homo-and hetero-polymers of acrylamide, are widely used as flocculants in the potable water industry, sewage treatment, paper manufacture and mining. However their utility is restricted and they cannot for instance be used in connection with foodstuffs because they are generally contaminated with unreacted acrylamide monomer which is a cumulative neurotoxin and a carcinogen. At present polyacrylamides are generally allowed in USA to contain up to 500 ppm of unreacted acrylamide and this limit is likely to be introduced elsewhere. This level of unreacted monomer can be achieved by use of longer polymerisation times and by a "heat treating" process. Such treatments increase cost of manufacture and reduce the efficiency of the polymers produced by causing branching in these polymers. Linear polymers have a high flocculation efficiency and are preferred. There is a need for a reliable process for reducing the level of unreacted monomer in polyacrylamides without damage to the polymer to 500 ppm and preferably to even lower levels (5-10 ppm or lower). If the unreacted monomer present could be reduced reliably to very low levels (1-5 ppm or less) polyacrylamides could be considered for uses from which they are presently excluded, e.g. in the manufacture of food contact materials or in direct contact with food.

Japanese Patent Publication No. 79011389 (corresponding to Japanese kokai No. 53086078) discloses a process for the decomposition of acrylamide monomer, e.g. in waste water or in polyacrylamides by contacting with an intracellular enzyme of Brevibacterium ammoniogenes preferably of strains ATCC 1641, ATCC 6871 and ATCC 6872. However, despite being known for a number of years, this process does not appear to have been used commercially to any significant extent. Moreover Brevibacterium ammoniogenes enzyme works poorly in polyacrylamide latex systems.

25 According to a first aspect of the present invention we provide a process for the decomposition of acrylamide in a medium containing it in which the medium is contacted with an enzyme capable of decomposing acrylamide under conditions suitable for the enzyme to decompose the acrylamide wherein the enzyme is an amidase enzyme which has been induced in a bacterium belonging to the species Methylophilus methylotrophus.

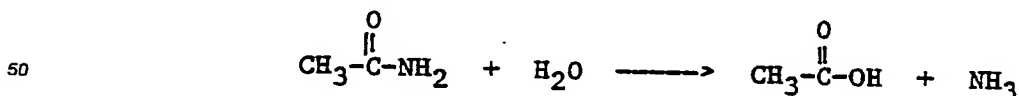
30 Further according to the invention we provide a method for the production of an amidase enzyme wherein a bacterium belonging to the species Methylophilus methylotrophus is cultivated in a medium containing appropriate nutrients and an amide under conditions such that the amidase enzyme is induced in the bacterium. The modified bacterium produced in the method of the invention is also included in the scope of the invention.

35 Further according to the invention we provide a process for the production of acrylic acid or a salt or ester thereof in which a medium containing acrylamide is contacted with an enzyme capable of decomposing acrylamide to acrylic acid under conditions suitable for the enzyme to decompose the acrylamide to acrylic acid or a salt or ester thereof wherein the enzyme is an amidase enzyme which has been induced in a bacterium belonging to the species Methylophilus methylotrophus.

40 Preferably the method of the invention is used to induce an amidase enzyme capable of decomposing acrylamide in the bacterium which enzyme may be used in the processes of the invention. Hereinafter the invention will be described mainly in terms of the production and use of this enzyme.

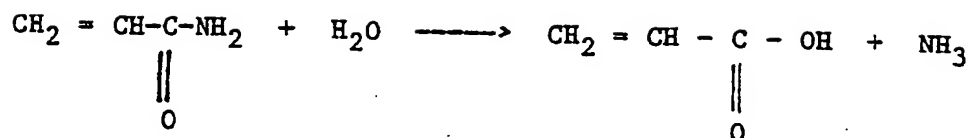
Amidases are part of the nitrogen assimilation apparatus of many cells grown under conditions where the sole source of nitrogen is an aliphatic amide.

45 Thus the metabolic rate of amidase is to release ammonia from said amide, e.g. by the following reaction:-



The ammonia released is then assimilated into protein biosynthesis. The aliphatic acid produced may or may not be assimilated dependent upon the microorganism type.

Amidases can also catalyse the conversion of acrylamide to acrylic acid by the following reaction:-



In the presence of the appropriate salts or alcohols the acrylic acid produced can be converted to its salt or ester.

The decomposition process of the invention can be used to decompose acrylamide in any medium in which it occurs. It is particularly useful for decomposing unreacted acrylamide present in polyacrylamide polymers and acrylamide present in waste waters, e.g. waste water from a process for the production of polyacrylamides. Polyacrylamides are produced as polymers of three types i.e. solution, dry and suspension polymers.

Polyacrylamides are produced as polymers of three chemical types. These are cationic, anionic and nonionic polymers wherein acrylamides may be copolymerised with other monomers, e.g. acrylic acid. These polymers may be manufactured by one of the following three basic technologies:-

Solution polymerisation wherein monomers are polymerised in aqueous solution to produce a gel-type product;

Dry polymers which are polymers produced as above but which are subsequently heat dehydrated prior to use;

Latex or suspension polymerisation wherein a solution of monomers in water is admixed with detergents and a non-aqueous low odour paraffinic solvent to form a stable suspension of aqueous droplets within which beads of polymers are formed by addition of a water soluble polymerisation initiator to the system.

Latex or suspension polymers form the largest group of polyacrylamides in terms of market share.

Anionic and neutral polyacrylamide polymers are readily treated by direct addition of amidase using the process of the invention or other processes, e.g. the process of our corresponding British Patent Application No. 8630012 (ICI Case B 34139) since they are formulated to have pH values in the range 6 to 9. Cationic polyacrylamide polymers may also be treated with amidase by the process of the invention or by other processes but these polymers are formulated to have acid pH values in the range 3 to 4 and treatment by direct addition of amidase is less effective than in the case of anionic neutral polymers.

According to a second aspect of the present invention we provide a process for the treatment of cationic polyacrylamide polymers to decompose unreacted acrylamide present therein using an amidase enzyme wherein the process comprises the following steps:-

(A) raising the pH of the polymer to be treated from a value in the range 3 to 3.5 to a value in the range 4 to 9; and thereafter

(B) treating the polymer with the enzyme under conditions suitable for the enzyme to decompose the acrylamide.

The process of the second aspect of the invention can be carried out using amidase from any source although amidase induced in strains of Methylophilus methylotrophus is preferred. The preferred values for other parameters described in this specification also apply to the process of the second aspect of the invention. In step (A) the pH may be raised by any suitable means. Preferably the pH is raised to a value in the range 4.5 to 7 and particularly to a value in the range 4.5 to 5.5 or 6.

The amidase enzyme used in the decomposition and acrylic acid production processes of the invention may be present in any suitable form in whole microorganism cells or as a crude or purified enzyme extract. The enzyme can be introduced to the processes in whole cells in the culture produced initially in the method of the invention or in a medium produced after only partial separation of water and other components from such a culture.

Generally however it is preferred that the cells should be separated from the culture before being used in the processes.

The species Methylophilus methylotrophus (formerly named Pseudomonas methylotrophus), strains of which are cultivated in the method of the invention, is described in our UK Patent Specification No. 1370892. Strains of this species suitable for use in the method of the invention have been deposited at the following 3 culture collection from which cultures are available:-

1. The National Collection of Industrial Bacteria (NCIB), Torrey Research Station, PO Box 31, 135 Abbey Road, Aberdeen AB9 8DG, Scotland, UK;

2. The Agricultural Research Culture Collection (NRRL), 1815 North University Street, Peoria, Illinois 61604, USA; and

3. The Fermentation Research Institute (FRI), Agency of Industrial Science and Technology, Ministry of International Trade and Industry, 1 - 3 Higashi 1-Chome, Yatabe-machi, Tsukuba-gun, Ibaragi-ken, Japan.

The corresponding accession numbers assigned to the strains deposited to these collections are as follows:-

- 5 NCIB 10508 - 10515 & 10592 - 10596, all inclusive;
 NRRL B 5352 - B 5364 inclusive; and
 FRI 1215 - 1227 inclusive.

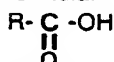
A preferred strain for use in the method of the invention is Methylophilus methylotrophus strain AS-1 (NCIB 10515) which may safely be used in treating, e.g. polyacrylamide polymers intended for use in connection with foods. The species Methylophilus methylotrophus and the above strains, particularly strain AS-1, have become widely known and are mentioned in numerous publications both by us and others. In addition to the deposits mentioned above cultures of them are also held in a number of University and other laboratories in the UK and in other countries. Also very suitable for use in the method of the invention is strain NCIB 11585, available from NCIB, the production of which by the genetic modification of strain NCIB 15 10515 is described in our European Patent Specification 35831 B and a number of other publications.

The amidase enzyme induced in Methylophilus methylotrophus has the properties set out below.

Methylophilus methylotrophus amidase properties

20	Physiological			
	status	Inducible		
	Location	Cytoplasmic		
	Protein	Soluble		
25		Acidic	Isoelectric point	4.2
		Tetrameric		
		Native MW	(Gel permeation)	155,000
30		Monomer MW	(SDS page)	38,000
	Properties	pH optimum	Acetamide	5
			Acrylamide	8
35		Temperature optimum		60°C
		Km	Acetamide	0.2 mM
			Acrylamide	10 mM
40		Activities	Acyl amidase	
			Acyl transferase	
45		Substrate	C ₁ <C ₂ <C ₃ >C ₄ >C ₅	
			Acetamide<Acrylamide>propionamide	

In the method of the invention a strain of Methylophilus methylotrophus is cultivated aerobically in a medium containing sources of carbon, nitrogen, phosphorus and other appropriate nutrients with an amide being present under conditions such that amidase is induced in the bacterial cells. Suitably cultivation takes place at a temperature in the range 20° to 40°C, preferably 34° to 38°C, and at a pH in the range 5.0 to 8.0, preferably 5.8 to 7.6. Cultivation can be by batch culture, single stage continuous culture or multiple stage continuous culture. A suitable dilution rate for continuous culture is in the range 0.05 hr⁻¹ to 0.55 hr⁻¹. Methanol is the preferred carbon source. Any suitable amide may be present in the culture medium. Acetamide or formamide are preferred. Other suitable amides include amide of carboxylic acids having the formula:-



Where R is a short chain aliphatic group, i.e. containing 1 to 5 carbon atoms, which may be a straight or a

branched chain. Preferably the amide is the sole or major nitrogen source and cultivation is by continuous culture under nitrogen limitation. A very suitable culture medium has the following composition:-

5	<u>Component</u>	<u>Amount present/litre</u>
	Phosphoric acid :	1.6 ml
	MgSO ₄ .7H ₂ O :	1.912 g
	K ₂ SO ₄ :	0.952 g
10	CuSO ₄ .5H ₂ O :	0.840 mg
	ZnSO ₄ .H ₂ O :	2.568 mg
	MnSO ₄ .4H ₂ O :	4.04 mg
15	FeSO ₄ .7H ₂ O :	37.20 mg
	Calcium formate :	0.173 g

20 Nitrogen is supplied either from acetamide alone or from acetamide and ammonia. Cells are grown in nitrogen excess or nitrogen limitation at a range of cell concentrations.

Cells produced by continuous, batch or fed batch fermentation can be harvested by any suitable means preferably by ultrafiltration or centrifugation to produce a slurry having 10 -25% by weight dry solids. Suitably this slurry is broken, e.g. by several freeze-thaw cycles or by mechanical breakage in a bead mill or french pressure device. Cell debris may then be removed by centrifugation leaving a crude cell-free 25 extract containing amidase. Heat treatment may take place before or after this centrifugation but preferably before as this gives improved sedimentation of the product during centrifugation in addition to stimulating activity. This extract can if desired be further purified by anionic ion exchange chromatography and gel filtration. The cell free amidase preparations are suitably stored cool at 0°-10°C or as frozen solutions or as freeze dried preparations prior to hydration and use.

30 When cells or cell free extracts containing amidase produced by the method of the invention are heated to temperatures in the range 40° to 80°C, as described in our co-pending British Patent Application No. 8630012 (ICI Case B 34139), we have found that, most unusually, the amidase activity is irreversibly increased by 1.5 to 35 times dependent upon the enzyme preparation. This effect is greatest at 35 temperatures in the range 55° to 65°C, especially in the range 58° to 62°C, at pHs in the range 4 to 9, especially at pHs between 6 and 7 and at protein concentrations in the range 0.1 to 200 mg/ml. Under these conditions little significant denaturation of amidase takes place.

Heating suitably takes place for a period in the range 20 minutes to 10 hours, especially 30 minutes to 180 minutes in some instances. To produce the increased activity the cells or cell-free enzyme can be heated immediately after they have been produced by the method of the invention or at some later time. 40 The method of the invention can usefully be carried out either as described above or as a similar process having four steps, i.e. a first fermentation step in which cells having induced amidase are grown in a fermenter, a second centrifugation step wherein cells are concentrated to a 10 to 20% dry solids slurry, a third heat shock step in which cells are subjected to heating to a temperature in the ranges described above and a fourth separation step in which a precipitate containing debris is separated from a supernatant 45 liquid containing the enzyme.

In the process of the invention amidase can be used to reduce free acrylamide residues occurring in all types of acrylamide polymers, particularly latices of the three basic chemical types. When treating a latex an amidase solution is added to the latex and is dispersed through the latex by stirring or similar mixing techniques at a level between 1 and 10,000 units/kg latex preferably 100 - 2000 U/kg latex. Incubation of 50 amidase with the latex results in conversion of free acrylamide to acrylic acid or a salt thereof. Incubation temperatures of 10°C to 100°C (particularly 30° to 80°C) are preferred. The latices are suitably treated over a pH range 3 to 10, preferably 5 to 7.

The amidase produced in the method of the invention converts acrylamide into acrylic acid. This reaction can therefore be used as a means for producing acrylic acid or, when carried out in the presence 55 of a suitable salt or alcohol, to produce salts or esters of acrylic acid.

The amidase produced by the method of the invention particularly when heated as described above exhibits very high activity. Thus use of the process of the invention for decomposing acrylamide greatly extends the range of utility of polyacrylamide polymers.

The invention is illustrated by the following Examples:-

EXAMPLE 1

Heat stimulation of the amidase activity of cell free extracts of *Methylophilus methylotrophus* AS-1

A series of cell free extracts prepared from *Methylophilus methylotrophus* AS-1 in a citrate-phosphate buffer system at pH 6.0 and at a concentration of 30 mg protein per ml were heated in a water bath for periods of 1 and 2 hours at a series of temperatures between 30° and 70°C. Samples were withdrawn at various times and were assayed for amidase activity at 30°C. The substrate used in all cases was acrylamide at a concentration of 100 mM. The assays were performed at 30°C in a 0.1 M citrate: 0.2 M phosphate buffer system at pH 6.0.

The results are shown in the Table and show increased activity under most of the conditions used with the highest increases being at 60°C. Increases were greater after 2 hours heating than after 1 hour.

Table

Treatment	Amidase Activity (U/ml)*		Stimulation	
none	3.69	-		
heating for 1 hr @ 30°C	3.69	1	x	
50°C	5.91	1.6	x	
60°C	24.65	6.7	x	
70°C	14.48	3.9	x	
heating for 2 hrs @ 30°C	3.76	1.02	x	
50°C	8.86	2.04	x	
60°C	40.06	10.8	x	
70°C	2.28	0.62	x	

* 1 unit is defined as 1 μ mole of NH_3 released per minute.

EXAMPLE 2

Cell-free extracts as described in Example 1 were incubated at 55°C, 60°C and 65°C respectively. Samples were taken at various times and assayed for amidase activity as described in Example 1. The results are set out in Figure 1 which is a graph of amidase activity (units/ml) against activation time (mins) and show optimal stimulation of amidase activity at 60°C over a period of two hours. Incubation at lower temperatures results in slower activation. Incubation at higher temperatures, e.g. at 65°C, shows an initial stimulation of rate of activation compared to 60°C but a lower final amidase activity.

EXAMPLE 3

Removal of acrylamide from an anionic latex.

An anionic latex, Nalfloc type A 626 produced by the Nalfloc Co. of Cheshire, UK, pH adjusted to 5.9 at 60°C was treated by addition of an amidase solution produced by the method of the invention of 300 units per ml activity to a final level of 1000 units amidase per kg latex. Samples were withdrawn at varying times and were analysed for free acrylamide by high pressure liquid chromatography (HPLC).

The results are set out in Figure 2 which is a graph of free acrylamide (mM) against treatment time (mins) and show a reduction in free acrylamide level from 1.78 mM (126 ppm) to 0.02 mM (1.4 ppm) within 30 minutes.

EXAMPLE 4Removal of acrylamide from a cationic latex.

A cationic latex, Nalfloc type 4625-SC, pH adjusted to 6.0 was treated as described in Example 3 with samples being analysed by HPLC.

The results are set out in Figure 3 which is a graph of free acrylamide (mM) against treatment time (mins) and show a reduction in free acrylamide level from 2.2 mM (156 ppm) to 0.02 mM (1.4 ppm) within 45 minutes.

EXAMPLE 5Removal of acrylamide from a non-ionic latex.

A non-ionic latex, Nalfloc type 8861-SC pH adjusted to 6.0 was treated as described in Example 3 with samples being analysed by HPLC.

The results are set out in Figure 4 which is a graph of free acrylamide (mM) against treatment time (mins) and show a reduction in free acrylamide level from 3.1 mM (220 ppm) to 0.15 mM (10.7 ppm) within 120 minutes.

EXAMPLE 6

An anionic latex Nalfloc type A.625 pH adjusted to pH 5.9 at 20°C was treated with amidase at a dose of 1000 units (U)/Kg A.625 as described in Example 3.

The results are set out in Figure 5 which is a graph of residual acrylamide (ppm BOL) against treatment time (mins) and show a reduction in free acrylamide level from 383 ppm (5.4 mM) to 161 ppm (2.27 mM) within 390 minutes.

"Standard abbreviation for "based on total weight of latex".

EXAMPLE 7

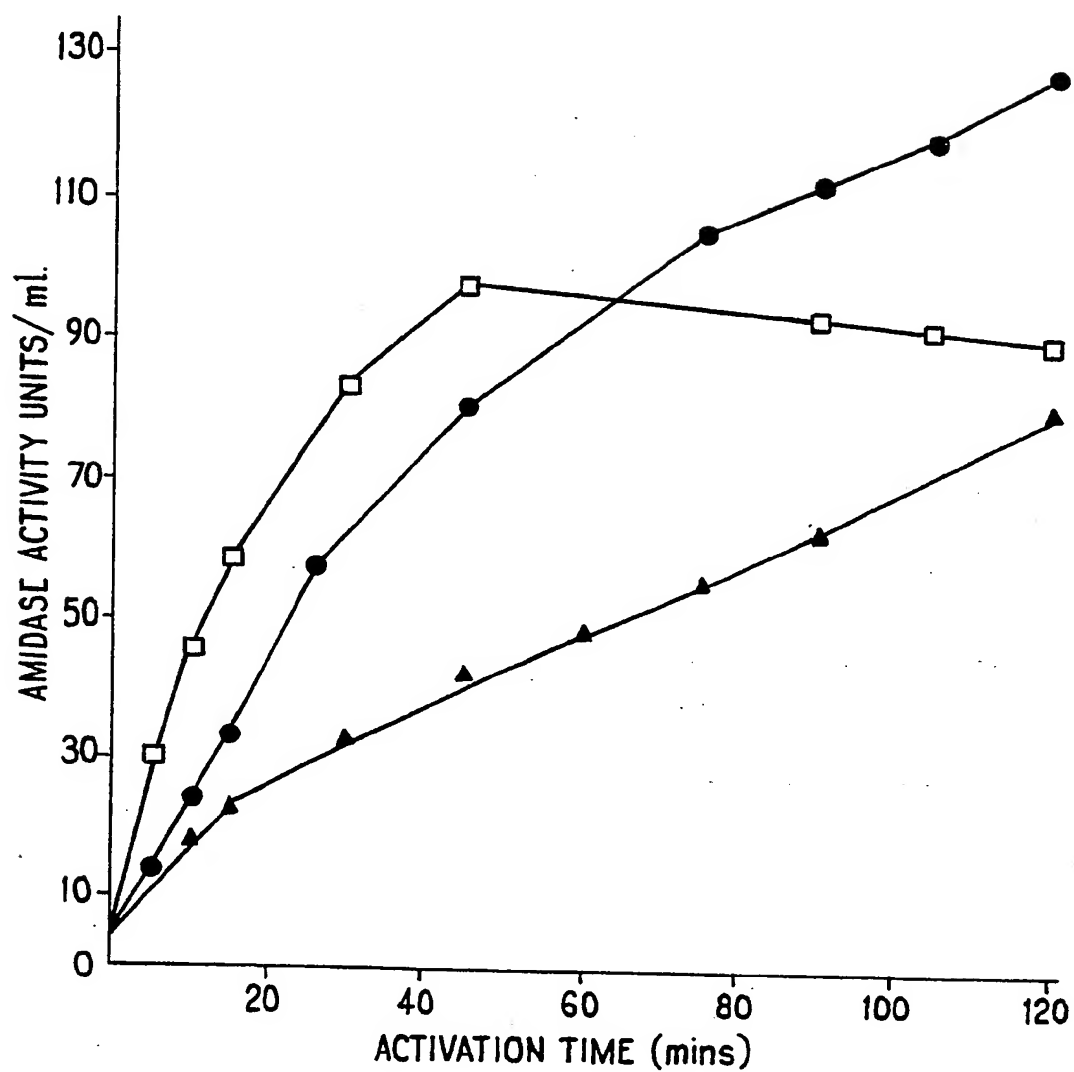
A cationic latex Nalfloc type 4625-SC was treated with amidase at a dose of 1000 units/Kg latex at 60°C at both its native pH (~4) and at an adjusted pH of 6. Samples were withdrawn at various times and assayed for acrylamide content by HPLC.

The results are set out in Figure 6 which is a graph of residual acrylamide ppm BOL against treatment time (mins) and show rapid reduction of free acrylamide in the latex adjusted to pH 6.0 but only limited reduction in the latex at its natural pH.

Claims

1. A process for the decomposition of acrylamide in a medium containing it in which the medium is contacted with an enzyme capable of decomposing acrylamide under conditions suitable for the enzyme to decompose the acrylamide wherein the enzyme is an amidase enzyme which has been induced in a bacterium belonging to the species Methylophilus methylotrophus.
2. A process according to claim 1 wherein the amidase enzyme has been induced in any of the strains NCIB 10508 to 10515, NCIB 10592 to 10596, all inclusive, and NCIB 11585 or variants or mutants of any of these strains.
3. A process according to claim 1 or claim 2 wherein the amidase enzyme has been heated to a temperature in the range 40° to 80°C for a sufficient period under suitable conditions to increase the activity of the enzyme.
4. A process according to claim 3 wherein the amidase enzyme has been heated to a temperature in the range 58° to 62°C.
5. A process according to claim 3 wherein the enzyme has been heated to a temperature in the range 40° to 80° whilst it is contained in broken cells and/or is associated with cell debris.
6. A method for the production of an amidase enzyme wherein a bacterium belonging to the species Methylophilus methylotrophus is cultivated in a medium containing appropriate nutrients and an amide under conditions such that the amidase enzyme is induced in the bacterium.
7. A method according to claim 6 wherein the amide is acetamide.
8. A method according to claim 6 or claim 7 wherein, after the amidase enzyme has been induced in the bacterium, bacterial cells are harvested and are then broken and the enzyme is heated to a temperature in the range of 40° to 80° whilst it is contained in broken cells and/or is associated with cell debris.
9. A process for the production of acrylic acid or a salt or ester thereof in which a medium containing acrylamide is contacted with an enzyme capable of decomposing acrylamide to acrylic acid under conditions suitable for the enzyme to decompose the acrylamide to acrylic acid or a salt or ester thereof wherein the enzyme is an amidase enzyme which has been induced in a bacterium belonging to the species Methylophilus methylotrophus.
10. A process for the treatment of cationic polyacrylamide polymers to decompose unreacted acrylamide present therein using an amidase enzyme wherein the process comprises the following steps:-
 - (A) raising the pH of the polymer to be treated from a value in the range 3 to 3.5 to a value in the range 4 to 9; and thereafter
 - (B) treating the polymer with the enzyme under conditions suitable for the enzyme to decompose the acrylamide.

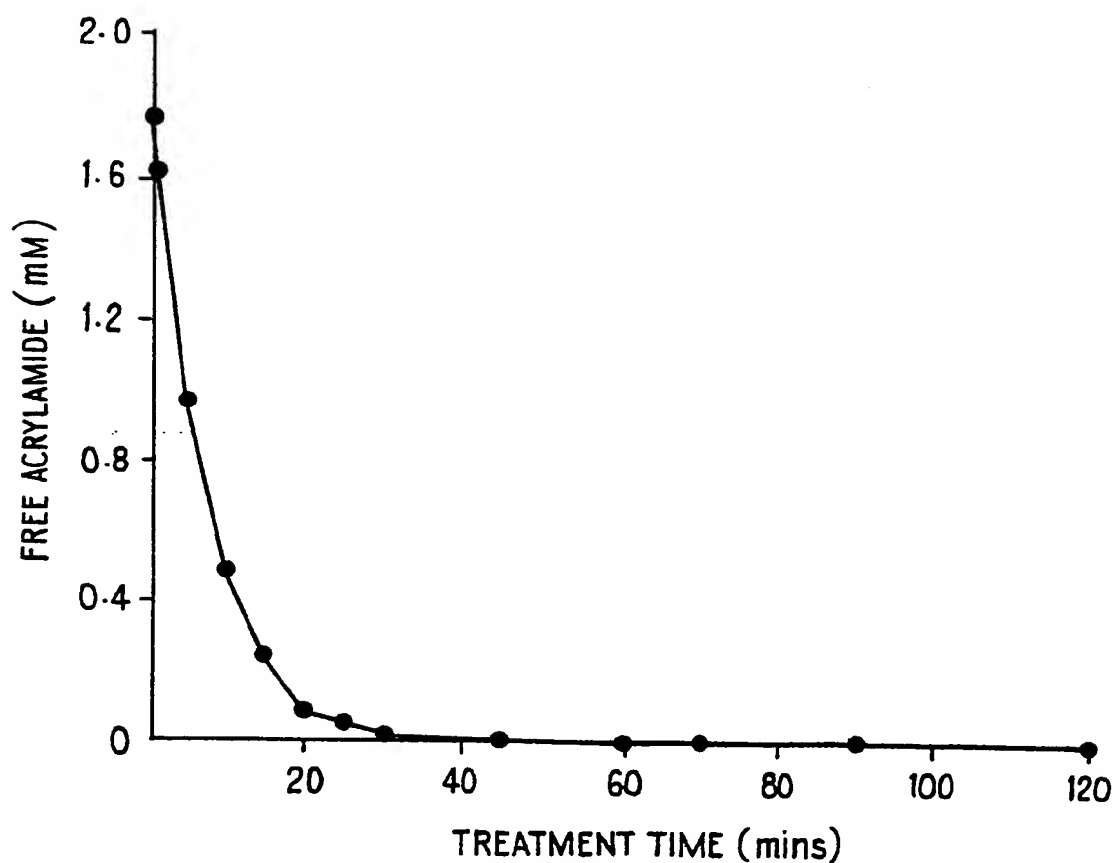
FIG.1

AMIDASE. HEAT ACTIVATION

- ● ACTIVATION TEMP. = 60°C
▲ ▲ " " 55°C
□ □ " " 65°C

FIG.2

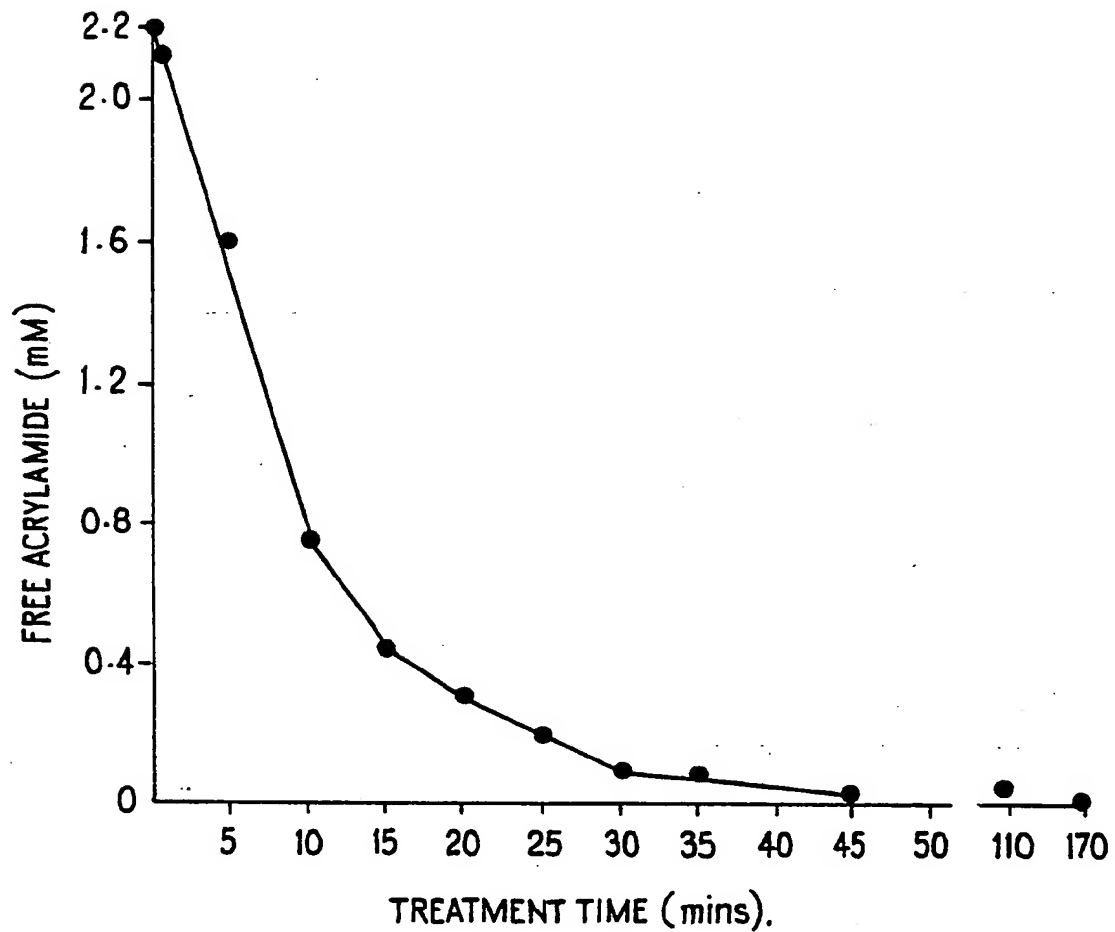
AMIDASE. REMOVAL OF ACRYLAMIDE FROM AN ANIONIC
LATEX TYPE A626



ANIONIC LATEX . NALFLOC TYPE A626
ENZYME LOADING 1000U/Kg
TEMPERATURE 60°C
pH 5.9.

FIG.3

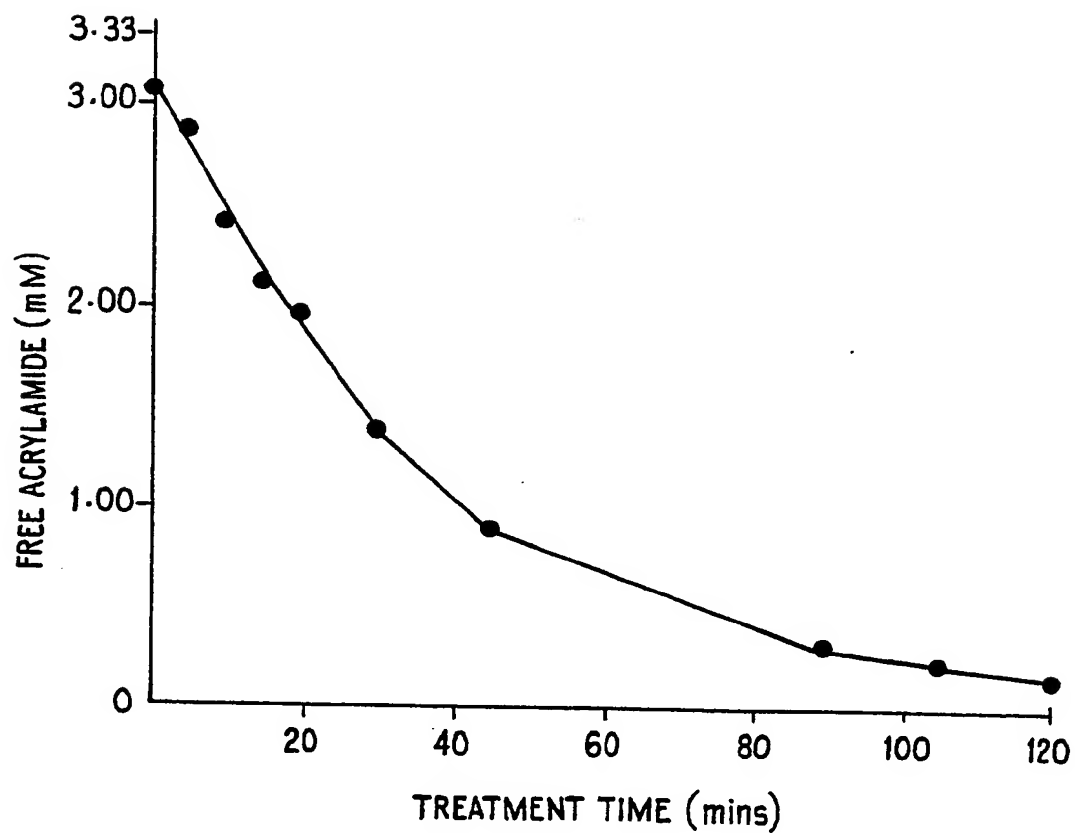
AMIDASE. REMOVAL OF ACRYLAMIDE FROM A CATIONIC LATEX
TYPE 4625 - SC.



CATIONIC LATEX. NALFLOC TYPE 4625-SC
ENZYME LOADING 1000 U/Kg LATEX
pH 6.0

FIG.4

REMOVAL OF FREE ACRYLAMIDE FROM A NONIONIC LATEX
NALFLOC TYPE 8861-SC.



TEMPERATURE 60°C

pH 6.0

ENZYME LOADING 1000 U/Kg LATEX.

FIG.5
AMIDASE TREATMENT OF ANIONIC LATEX A625

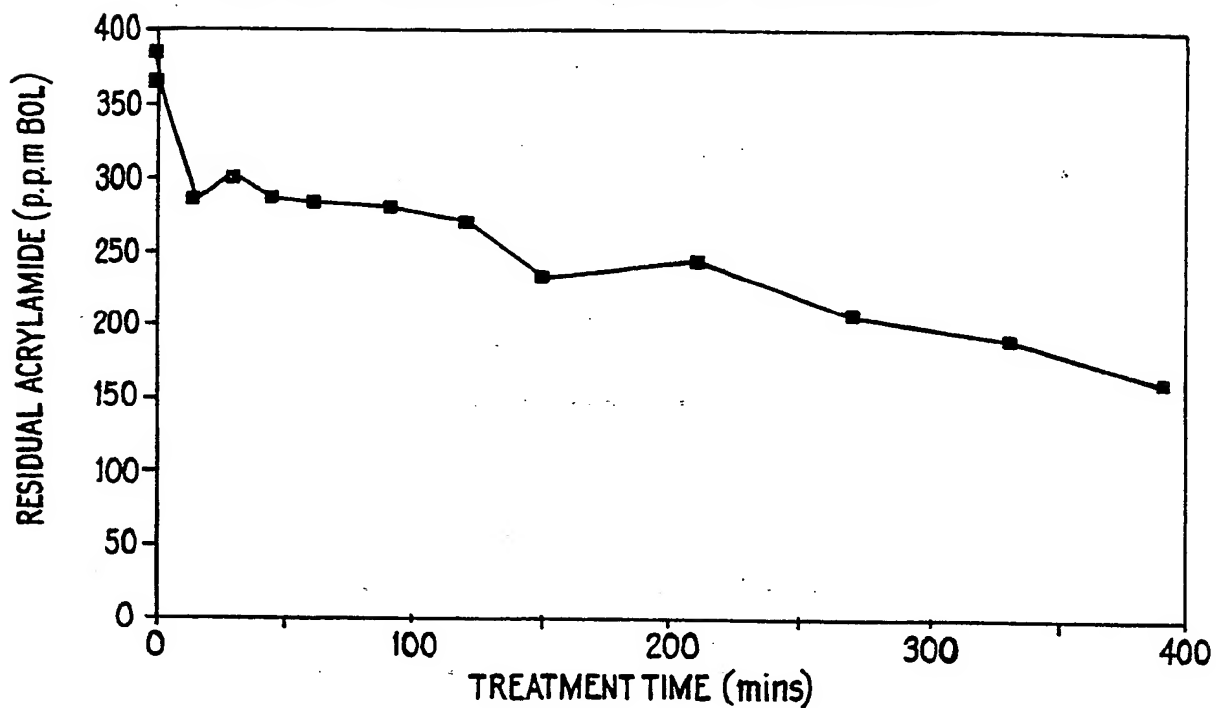
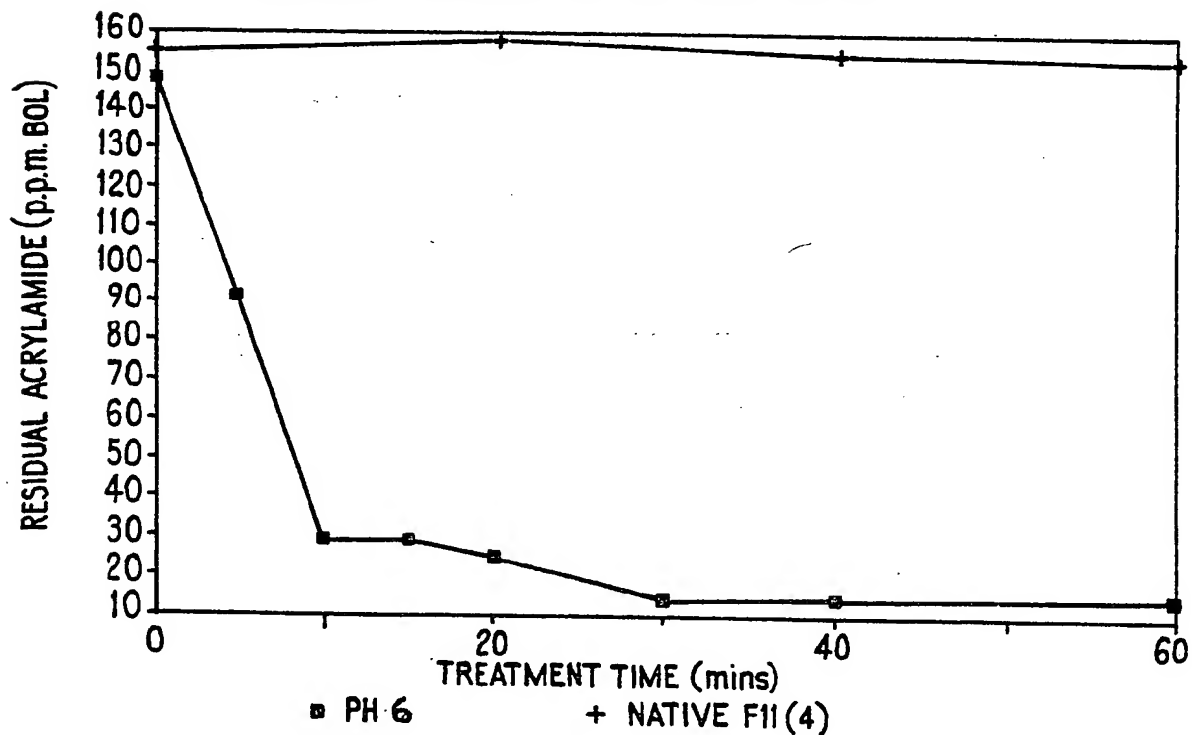


FIG.6
AMIDASE TREATMENT OF CATIONIC LATEX



THIS PAGE BLANK (USPTO)

12 **EUROPEAN PATENT APPLICATION**

21 Application number: 87310709.8

51 Int. Cl.⁵: C12N 9/80, C12P 7/40,
 //(C12N9/80,C12R1:01)

22 Date of filing: 04.12.87

30 Priority: 16.12.86 GB 8630029

43 Date of publication of application:
 22.06.88 Bulletin 88/25

84 Designated Contracting States:
 BE DE ES FR GB IT NL

98 Date of deferred publication of the search report:
 16.05.90 Bulletin 90/20

71 Applicant: **IMPERIAL CHEMICAL INDUSTRIES PLC**
 Imperial Chemical House, Millbank
 London SW1P 3JF(GB)

72 Inventor: **Byrom, David**
 27 Moor Park Nunthorpe
 Middlesborough Cleveland TS7 0JJ(GB)
 Inventor: **Carver, Mark Andrew Hallswood**
 503 Yarm Road Eaglescliffe
 Stockton-on-Tees Cleveland TS16 9BE(GB)

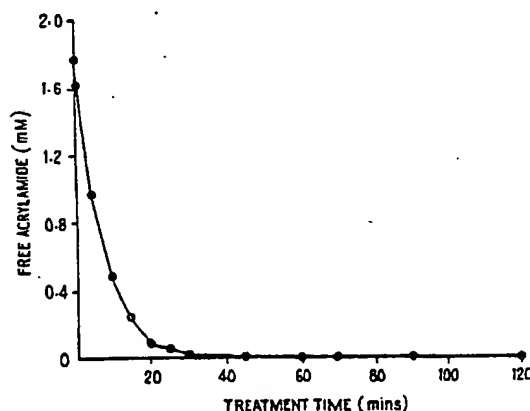
74 Representative: **Aufflick, James Neil et al**
 Imperial Chemical Industries PLC Legal
 Department: Patents PO Box 6 Bessemer
 Road
 Welwyn Garden City Herts, AL7 1HD(GB)

54 **Process for the decomposition of acrylamide.**

57 A process for the decomposition of acrylamide using an amidase (acrylamide amido hydrolase) enzyme in which the enzyme is induced in strains of *Methylophilus methylotrophus*. A method for inducing the enzyme and a process for producing acrylic acid are also claimed. The process for the decomposition of acrylamide is useful in particular for reducing the level of unreacted monomer associated with homo- and hetero- polymers of acrylamide. The activity of the enzyme can be increased by heating it to a temperature in the range 40° to 80° C.

FIG.2

AMIDASE REMOVAL OF ACRYLAMIDE FROM AN ANIONIC LATEX TYPE A626



ANIONIC LATEX, NALFLOC TYPE A626
 ENZYME LOADING 1000U/Kg
 TEMPERATURE 60°C
 pH 5.9

EP 0 272 025 A3



DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int. Cl.4)
A	CHEMICAL ABSTRACTS, vol. 96, no. 25, 21st June 1982, page 396, abstract no. 214089s, Columbus, Ohio, US; T. ARAI et al.: "Biodegradation of acrylamide monomer by a Rhodococcus strain", & ZENTRALBL. BAKTERIOL., MIKROBIOL. HYG., ABT. 1, SUPPL. 1981, 11 (Actinomycetes), 297-307 ---		C 12 N 9/80 C 12 P 7/40 // (C 12 N 9/80 C 12 R 1:01)
A	CHEMICAL ABSTRACTS, vol. 91, no. 1, 7th January 1980, page 610, abstract no. 89550d, Columbus, Ohio, US; & JP-A-79 46 887 (NITTO CHEMICAL INDUSTRY CO., LTD.) 13-04-1979 ---		
A	CHEMICAL ABSTRACTS, vol. 93, no. 5, 4th August 1980, page 709, abstract no. 43912m, Columbus, Ohio, US; R.K. DALAL et al.: "A microbial route for acrylic acid production", & BIOSOURCES DIG. 1980, 2(2), 89-97 ---		
A	AGRIC. BIOL. CHEM. vol. 46, no. 5, 1982, pages 1175-1181, Tokyo, JP; Y. ASANO et al.: "Purification and characterization of amidase which participates in nitrile degradation" ---		
A	EP-A-0 093 910 (BASF AG) ---		
E	EP-A-0 272 026 (ICI) * Whole document * -----	1-10	
The present search report has been drawn up for all claims			
Place of search THE HAGUE		Date of completion of the search 20-02-1990	Examiner VAN PUTTEN A.J.
CATEGORY OF CITED DOCUMENTS X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons ----- & : member of the same patent family, corresponding document			

EPO FORM 1503 03.82 (P0401)